



CrossMark  
click for updates

Cite this: *RSC Adv.*, 2015, 5, 70339

# Exploring the potential of exfoliated ternary ultrathin $\text{Ti}_4\text{AlN}_3$ nanosheets for fabricating hybrid patterned polymer brushes†

Qun Ye,<sup>‡a</sup> Peng Xiao,<sup>‡a</sup> Wulong Liu,<sup>a</sup> Ke Chen,<sup>a</sup> Tao Chen,<sup>\*a</sup> Jianming Xue,<sup>b</sup> Shiyu Du<sup>a</sup> and Qing Huang<sup>\*a</sup>

Since the discovery of graphene, two-dimensional (2D) materials have been receiving increased attention. The quest for new 2D materials with unique structure and special properties has become urgent. Herein we report on the preparation of a new kind of ternary 2D material,  $\text{Ti}_4\text{AlN}_3$  nanosheets, by liquid exfoliation of the corresponding laminated MAX phase. The obtained  $\text{Ti}_4\text{AlN}_3$  nanosheets, bearing abundant surface groups, can be further used to fabricate micro-patterns *via* micro-contact printing ( $\mu\text{CP}$ ) and subsequently functionalized through self-initiated photografting and photopolymerization (SIPGP) to achieve MAX-based hybrid patterned polymer brushes. Our work opens a door to explore the synthesis of 2D hybrid materials for functional applications based on the traditional MAX phases.

Received 18th May 2015  
Accepted 12th August 2015

DOI: 10.1039/c5ra09227b

[www.rsc.org/advances](http://www.rsc.org/advances)

## Introduction

Two-dimensional (2D) materials have garnered tremendous interest recently. They offer high specific surface area, diverse electronic structure and properties compared with their bulk counterparts. Among typical 2D materials, graphene is arguably the most studied due to its unique mechanical, thermal, electronic, and optical properties.<sup>1</sup> Other 2D materials containing two elements, such as hexagonal boron nitride (h-BN),<sup>2,3</sup> transition metal dichalcogenides,<sup>4</sup> and metal oxides,<sup>5</sup> have also been receiving increased attention. Recently, a new family of binary 2D materials, named MXenes, was reported.<sup>6</sup> They were produced by etching of A elements from MAX, or  $\text{M}_{n+1}\text{AX}_n$  phases (M is an early transition metal, A is mainly an A-group element, X is carbon and/or nitrogen, and  $n = 1, 2, \text{ or } 3$ )<sup>7–9</sup> by F-containing etchants (HF,  $\text{NH}_4\text{HF}_2$ ,<sup>10</sup> or the mixture<sup>11</sup> of fluoride salts and hydrochloric acid), leaving  $\text{M}_{n+1}\text{X}_n$  layers. So far the MXene family includes  $\text{Ti}_3\text{C}_2$ ,  $\text{Ti}_2\text{C}$ ,  $(\text{Ti}_{0.5}, \text{Nb}_{0.5})_2\text{C}$ ,  $(\text{V}_{0.5}, \text{Cr}_{0.5})_3\text{C}_2$ ,  $\text{Ti}_3\text{CN}$ ,  $\text{Ta}_4\text{C}_3$ ,  $\text{Nb}_2\text{C}$ ,  $\text{V}_2\text{C}$ , and  $\text{Nb}_4\text{C}_3$ .<sup>12–15</sup> These MXenes have greatly enlarged the number of binary 2D materials, and showed promising performance as electrode materials in both lithium ion batteries (LIBs) and

supercapacitors,<sup>14,16,17</sup> catalyst supports in the chemical industry,<sup>18,19</sup> and adsorbents for heavy metal ions.<sup>20</sup>

These unexpected 2D materials, MXenes derived from MAX phases intrigue the exploration of multi-element graphene-like materials. In fact, in the  $\text{Ti}_3\text{Si}_{0.75}\text{Al}_{0.25}\text{C}_2$  (TSAC) case, the partial solid solution of Al in Si site weakened the bond strength of A layers, and favored the formation of TSAC nanosheets.<sup>21</sup> Compared with ones comprising of single or binary element, these multi-element 2D materials broaden the opportunity to tune their physical and chemical properties. Moreover, the above-mentioned discovery of MXene materials only takes advantage of features of their binary MX units, not of parent ternary MAX phase, whose appealing structural and functional characters, such as excellent conductivity, good oxidation resistant,<sup>22</sup> and radiation tolerance,<sup>23</sup> have been widely studied.<sup>7</sup> Hence, an effective strategy to exfoliate the bulk MAX phases into 2D ones with intact crystal configuration for functional applications is highly needed.

Inspired by the exfoliation behavior of carbide MAX phase to 2D MXene<sup>6</sup> or MAX phase nanosheets,<sup>21</sup> herein, we report a strategy to prepare exfoliated 2D nitride MAX nanosheets (using  $\text{Ti}_4\text{AlN}_3$  as a sample) derived from the corresponding MAX phase by etching in diluted hydrofluoric acid (HF) with a following ultrasonic treatment. Unlike TSAC sheets<sup>21</sup> directly exfoliated from A-layer-activated MAX phase with ultrasound more than 12 hours, the  $\text{Ti}_4\text{AlN}_3$  powders are firstly etched in the caustic HF solution with a suitable concentration. Many defects form in the structure due to the corrosion, and the following exfoliation process of  $\text{Ti}_4\text{AlN}_3$  powders to nanosheets becomes easier and more efficient. In order to further explore the potential of this new 2D inorganic material, micro-patterns and chemical functionality of  $\text{Ti}_4\text{AlN}_3$  nanosheets were

<sup>a</sup>Ningbo Institute of Material Technology and Engineering, Chinese Academy of Science, Ningbo, People's Republic of China. E-mail: tao.chen@nimte.ac.cn; huangqing@nimte.ac.cn

<sup>b</sup>State Key Laboratory of Nuclear Physics and Technology, Peking University, Beijing, 100871, People's Republic of China

† Electronic supplementary information (ESI) available: Details of experimental procedures and additional data. See DOI: 10.1039/c5ra09227b

‡ Qun Ye and Peng Xiao contribute equally.

attempted. After etching in HF solution, the surface of materials is usually attached by hydrophilic hydroxyl and fluorine groups.<sup>6,14</sup> As an active site, -OH group has been successfully utilized to fabricate micro-patterned graphite oxide (GO) films through micro-contact printing ( $\mu$ CP) induced supramolecular self-assembly of GO nanosheets,<sup>24</sup> and initiated to grow poly(2-(dimethylamino)ethyl methacrylate) (PDMAEMA) polymer brushes *via* self-initiated photografting and photopolymerization (SIPGP) from  $V_2C$  MXene to achieve the “smart” properties of  $CO_2$  and temperature dual response.<sup>25</sup> The fabrication of hybrid polymer brushes with controllable sizes and special properties can vastly extend the applications of the corresponding organic or inorganic material. Our work thus allows us to achieve 2D hybrid patterned polymer brushes grafted from MAX nanosheets, which will endow the traditional MAX phases to have more smart functionality.

## Experimental

### Materials

$Ti_4AlN_3$  phase used in this work was synthesized from the starting raw materials of Ti, Al, and TiN powders with a non-stoichiometric molar ratio of 1 : 1.2 : 2.05 by a microwave sintering method at 1250 °C for 30 min in an argon atmosphere.<sup>26</sup> Styrene (~99% purity, Alfa Aesar China (Tianjin) Co., Ltd) was purified by neutral  $Al_2O_3$  column chromatography and dried with a 0.4 nm molecular sieve at room temperature for 3 days. Patterned PDMS stamps were fabricated from Sylgard 184 (the ratio between component A and B was 1 : 10) on a silicon master. Silicon wafers were cleaned in a mixture of  $H_2O_2/H_2SO_4$ , 1 : 3 (v/v, *i.e.* “piranha solution”) at 80 °C for 2 h and washed thoroughly with Milli-Q-grade water.

### Preparation of $Ti_4AlN_3$ ink

Roughly 3 g  $Ti_4AlN_3$  powders were immersed in about 50 mL of 20 wt% diluted  $HF_3$  solution at room temperature for 30 h. The etched powders were collected by centrifugal treatment and then washed several times using deionized water. To obtain stable suspension, the powders were then dispersed into alcohol and ultrasonically treated for nearly 6 h. After standing still for more than 24 h, the solution was centrifuged at about 3000 rpm to remove large-size particles, leaving a stable ink.

### Micro-contact printing ( $\mu$ CP)

The patterned PDMS stamp was inked by exposing the stamp features to the ink of  $Ti_4AlN_3$  sheets for 3 min and drying with nitrogen, and then brought into contact with Si substrate for 1 min with an adequate applied force of 50 g to fabricate the patterns on the substrate.

### Self-initiated photografting and photopolymerization<sup>27</sup>

The patterned Si substrates surface was submerged in 2 mL of styrene monomer and irradiated with an UV fluorescent lamp with a spectral distribution between 300 and 400 nm ( $\lambda_{max} = 365$  nm with a total power of  $\sim 240$  mW  $cm^{-2}$ ) for 1 h in  $N_2$  atmosphere. After SIPGP, the functionalized substrate were

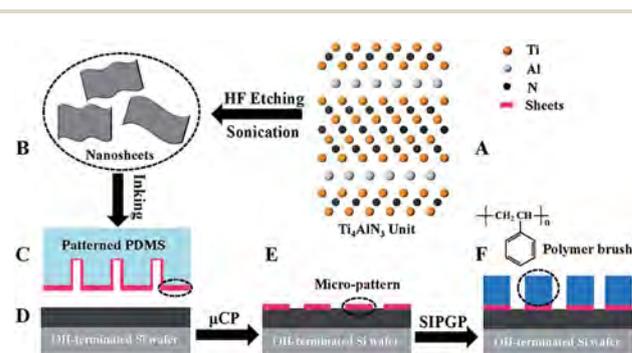
rinsed with several solvents (toluene, ethyl acetate, and ethanol for styrene) to remove physisorbed polymer.

### Characterization

The phase composition of samples before and after immersion was analyzed by X-ray diffraction (XRD), using a diffractometer (D8 Advance, Bruker AXS, Germany) with Cu  $K_\alpha$  radiation. Their microstructures were characterized by a scanning electron microscope (SEM, FEI Quanta 250 FEG, Hillsboro, OR), and transmission electron microscopy (TEM, FEI Tecnai F20) at an acceleration voltage of 200 kV. Atomic force microscopy (AFM) study in this work was taken by a multimode AFM (Being Nano-Instruments, Ltd) operating in the contact and/or tapping mode using silicon cantilevers.

## Results and discussion

The preparation process of the exfoliated ternary  $Ti_4AlN_3$  nanosheets and subsequent exploration for fabricating hybrid patterned polymer brushes is outlined in Scheme 1. To the best of our knowledge, it is the first effort to prepare nitride MAX nanosheets and MAX-based hybrid brushes. In order to find a suitable etching condition for exfoliation, different concentration of HF solution and etching time were initially attempted. In the diluted HF solution (*i.e.* 10 wt%), the powders seemed intact due to low reaction kinetics. However, in the concentrated HF solution (*i.e.* 40 wt%),  $Ti_4AlN_3$  was vigorously eroded, and impurities such as  $Al_2O_3$  and TiN always remained in the final products (Fig. S1 and S2†). In a moderate condition such as 20 wt% HF for 30 h, the accordion-like structure could be easily found in the final products (Fig. 1a and S3†). These etched products were well dispersed in water or alcohol after sonication, and formed a stable suspension even after standing still for more than 24 h (Fig. S4†). For comparison, pure  $Ti_4AlN_3$  powders without HF-etching treatment cannot form such suspension (Fig. S4†), which is the same as the result of ultrasonic-treated  $Ti_3SiC_2$ .<sup>21</sup>



Scheme 1 Schematic procedure of preparing nitride  $Ti_4AlN_3$  nanosheets, and subsequently fabricating hybrid patterned polymer brushes. (A and B) The process of etching  $Ti_4AlN_3$  powders in the diluted hydrofluoric acid (20 wt%) with a following ultrasonic treatment for 6 h. (C–E) Fabricating micro-patterns of  $Ti_4AlN_3$  nanosheets *via* micro-contact printing ( $\mu$ CP). (F) Growing polymer brushes through self-initiated photografting and photopolymerization (SIPGP) on the micro-patterns.

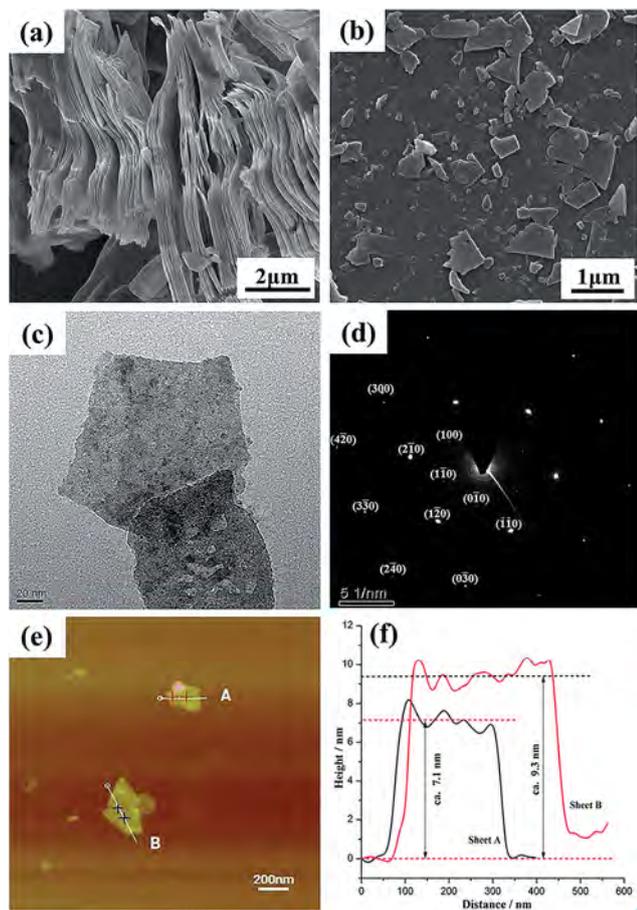


Fig. 1 (a) SEM image of as-etched  $\text{Ti}_4\text{AlN}_3$  particles in 20 wt% HF. (b) Typical morphology of etched  $\text{Ti}_4\text{AlN}_3$  powders after sonication for 6 h. (c) TEM image of as-obtained nanosheets, and (d) the corresponding SAED pattern. (e) AFM image of as-obtained nanosheets, and (f) the corresponding height measurement.

In order to find out whether there was a successful exfoliation of  $\text{Ti}_4\text{AlN}_3$  particles to nanosheets, scanning electronic microscopy (SEM) was employed to investigate the morphology of the products in the suspension. As shown in Fig. 1b, sheets with a wide range of diameters from several nanometers to micrometers are obtained after sonication, which resemble the typical delaminated  $\text{Ti}_3\text{C}_2$  MXene.<sup>6</sup> A transmission electron microscopy (TEM) analysis (Fig. 1c and d) of the sheets reveals the graphene-like products are still  $\text{Ti}_4\text{AlN}_3$  phase, which is not as-reported MXene material (*i.e.*  $\text{Ti}_4\text{N}_3$  phase) as  $\text{Ti}_3\text{AlC}_2$  phase normally does under the same etching condition. The corresponding selected area electron diffraction (SAED) pattern (Fig. 1d) clearly reveals the hexagonal crystalline nature of the sheet, and the corresponding lattice spacing of (100), (110), and (300) planes (listed in Fig. S5a†) are all consistent with those of  $\text{Ti}_4\text{AlN}_3$  phase. Energy dispersive spectroscopy (EDS) data (Fig. S5b†) shows the presence of Al as well as Ti and N elements, indicating the existence of  $\text{Ti}_4\text{AlN}_3$  phase, which further corroborates the SAED result. C and Cu signals originate from the supporting foil, and vast O, F should come from the surface groups of hydroxyl and fluorine groups, which have

been proved to exist in the MXene materials after HF-etching process.<sup>6,14</sup> To confirm the thickness of the exfoliated sheets, tapping-mode atomic force microscopy (AFM) analysis was employed. As shows in Fig. 1e and f, the thickness of the as-exfoliated nanosheets is only 7–9 nm, which is about three to four multilayers of the  $\text{Ti}_4\text{AlN}_3$  crystal unit along  $c$  axis. The sharp fluctuation of curves in Fig. 1f should be due to the existence of holes or small particles on the surface of sheets.

The formation of  $\text{Ti}_4\text{AlN}_3$  nanosheets is further confirmed by X-ray diffraction (XRD). As Fig. 2 shows, the major phase of the exfoliated sheets (curve C) drying on the silicon wafer is still  $\text{Ti}_4\text{AlN}_3$  phase (JCPDS: 65-9771), the same with the results of  $\text{Ti}_4\text{AlN}_3$  before (curve A) and after (curve B) HF-etching treatment. However, it clearly shows that the relative intensity of peaks belonging to (00 $l$ ) planes of  $\text{Ti}_4\text{AlN}_3$  crystal structure increases, but the peaks assigned to (104), (105), (106) lattice planes obviously weaken. This result indicates the samples lying flat on the silicon wafer expose more basal plane (00 $l$ ) to X-ray compared with the corresponding MAX phase powders, as SEM revealed (Fig. 1b). Furthermore, there is still small amount of TiN phase existing in the as-prepared  $\text{Ti}_4\text{AlN}_3$  sheets, most possibly attached on the surface of  $\text{Ti}_4\text{AlN}_3$  nanosheets as AFM technique detected. Raman spectroscopy (Fig. S6†) also confirms the  $\text{Ti}_4\text{AlN}_3$  phase in the products. All the characteristic Raman peaks<sup>28</sup> belonging to  $\text{Ti}_4\text{AlN}_3$  phase still remained after etching and sonication. The corresponding calculated peak energies and symmetries for the assigned modes are summarized in Table S1.†

At present, the mechanism for the formation of ternary 2D  $\text{Ti}_4\text{AlN}_3$  nanosheets (rather than  $\text{Ti}_4\text{N}_3$  MXene) is not exactly known, but it could be due to the dissolution of one complete  $\text{Ti}_4\text{AlN}_3$  unit layer (both Al and  $\text{Ti}_4\text{N}_3$  layers) rather than single Al atom layer in HF. Since the exfoliation of MAX phases is mainly due to the active A atom (Al in all current reports on MXenes),<sup>6,12–15</sup> the bond strength between A atom to M or X

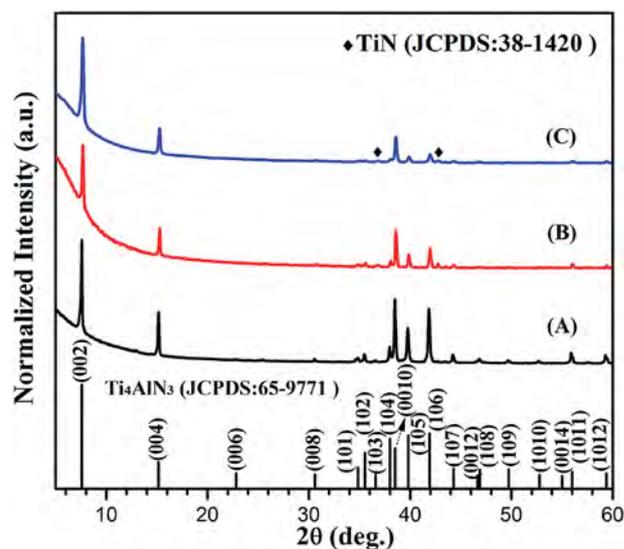


Fig. 2 XRD patterns for  $\text{Ti}_4\text{AlN}_3$  before (A) and after (B) 20 wt% HF treatment, and for the exfoliated sheets film (C).

atom is the decisive factor to control the etching behavior in HF solution. When the X atom changes from C to N, the exfoliation behavior of MAX phase is totally different. The stronger bond<sup>29</sup> between Al atoms to Ti–N substructure unites the two layers, and prevents the formation of  $TiN_x$  phase in HF solution. On the contrary, Al atom is much free in the  $Ti_3AlC_2$  phase, which well explains the formation of binary MXene materials. Furthermore,  $Ti_2AlN$  phase has the similar periodic lattice structure except the thinner Ti–N layer than  $Ti_4AlN_3$  phase. Through carefully selecting the etching condition (e.g. in 10 wt% HF for 10 hours),  $Ti_2AlN$  nanosheets can also be prepared in our work. Since N atom can be doped into or replace most of carbide MAX phase, it would be an effective strategy to synthesize 2D ternary MAX phase.

Recently, freestanding films have attracted continuing attention due to their potential applications in micro-sensors<sup>30</sup> or energy storage devices.<sup>31</sup> We found that the obtained  $Ti_4AlN_3$  ink could be successfully used to fabricate ultrathin film using Langmuir–Blodgett assembly,<sup>32</sup> as Fig. 3 showed. The as-fabricated Langmuir–Blodgett (LB) film attaching on the PDMS (Fig. 3a) possesses a good flexibility to bend with the substrate (Fig. 3b), and it seems semi-transparent. An AFM height measurement (Fig. 3d) of the film attaching on the silicon wafer (Fig. 3c) shows that the thickness of  $Ti_4AlN_3$  Langmuir–Blodgett film is only 120 nm, which is comparable to other functional films (e.g. CNT film<sup>33</sup>), and may have the potential in electronic or photonic applications.<sup>10</sup>

With the development of UV polymerization, SIPGP has provided a convenient method to grow polymer brushes. In fact, we have successfully functionalized another two-dimension material derived from MAX phase (i.e.  $V_2C$  MXene) with

PDMAEMA *via* SIPGP,<sup>25</sup> which endows  $V_2C$  to have tunable conductivity during  $CO_2$  uptake and subsequent  $CO_2$  release, and makes the hybrid  $V_2C$  as a smart material. Moreover, in order to meet the practical application, we have also fabricated micro-patterns of GO sheets *via*  $\mu$ CP.<sup>24</sup> The grid patterns with adjustable sizes and shapes are subsequently grafted polystyrene (PS) *via* SIPGP to achieve patterned polymer brushes, which play an important role in the modification of surface properties.<sup>34,35</sup> Hence, the functionalization of 2D materials is an effective way to achieve new structures and more properties.

For  $Ti_4AlN_3$  nanosheets, it has shown that they are bearing abundant hydroxyl groups as EDS (Fig. S5b†), FTIR (Fig. S8†) and XPS (Fig. S9†) results indicated. Based on the different hydrogen bond strengths of PDMS– $Ti_4AlN_3$  and  $Ti_4AlN_3$ –silicon interfaces, micro-patterns of  $Ti_4AlN_3$  sheets were easily transferred from PDMS stamp onto Si wafer *via*  $\mu$ CP, and formed patterns with a size of  $10\ \mu m \times 10\ \mu m$  and a thickness of 7 nm (Fig. 4a and b) on the substrate. Furthermore, as a photo active site, –OH group on the patterned sheets could be initiated to grow polymer brushes by self-initiated photografting and photopolymerization (SIPGP) from the micro-patterns without a surface bonded initiator.<sup>27</sup> Hence, after immersing the substrate in the styrene monomer solution and irradiating with a UV lamp for an hour, we can grow polystyrene (PS)<sup>24</sup> from  $Ti_4AlN_3$  nanosheets, and achieve functionalized MAX micro-patterns. After polymerization, the resulted thickness is about 160 nm (Fig. 4c and d), showing a drastic increase in height, which indicates the successful graft of PS.

In order to achieve more unique nanostructures with special properties, the graft of polymer brushes from MAX sheets cannot be limited to PS or  $Ti_4AlN_3$  phase selected in our work,

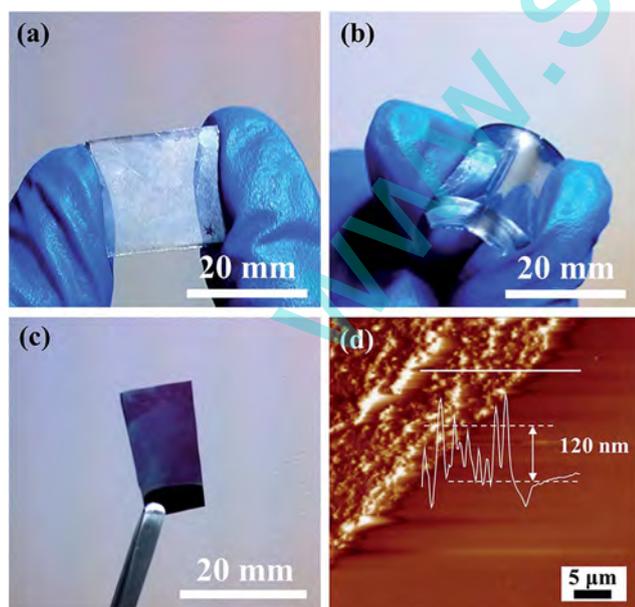


Fig. 3 (a and b) Digital photographs of Langmuir–Blodgett (LB) film of  $Ti_4AlN_3$  sheets attaching on the PDMS substrate. (c) Photograph of  $Ti_4AlN_3$  LB film attaching on a Si wafer, and (d) the corresponding AFM measurement of the film.

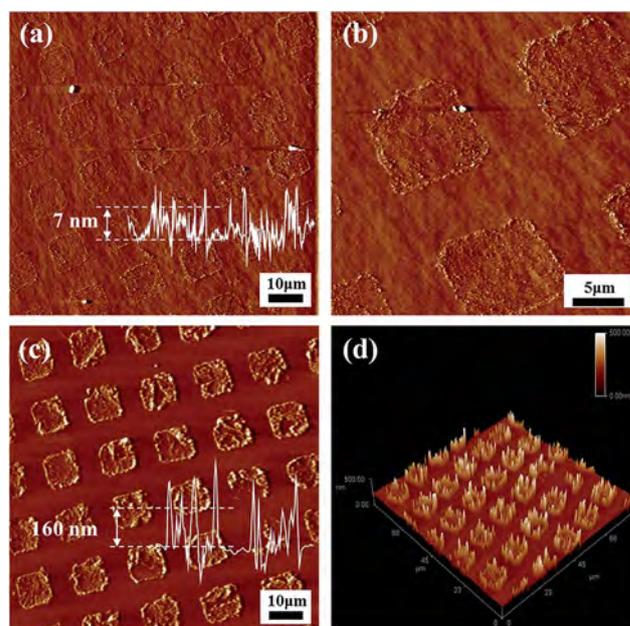


Fig. 4 (a) AFM height image of micro-patterned  $Ti_4AlN_3$  nanosheets, and (b) the corresponding amplified image. (c) AFM height image of polystyrene (PS) brushes grafted from micro-patterned  $Ti_4AlN_3$  nanosheets, and (d) the corresponding 3D view.

but should be extended to other polymers or 2D material. This work just opens a door to explore the synthesis of 2D hybrid materials for more functional applications<sup>25,36</sup> basing on the traditional MAX phases.

## Conclusions

In summary, a new kind of ternary 2D material, Ti<sub>4</sub>AlN<sub>3</sub> phase nanosheets was prepared for the first time through a simple soaking and ultrasonic treating approach. This finding shows that the replacement in the X atom from C to N in MAX phases can dramatically changes their exfoliation behavior in HF solution, and also implicates that more nitride 2D materials can be prepared from their corresponding MAX bulk counterpart. The obtained Ti<sub>4</sub>AlN<sub>3</sub> nanosheets containing vast surface groups can be further functionalized to fabricate hybrid patterned polymer brushes through micro-contact printing induced supramolecular self-assembly and subsequent self-initiated photografting and photopolymerization. This micro-patterned MAX-based hybrid system endows the traditional MAX phase to have chemical functionality and more potential applications.

## Conflict of interest

The authors declare no competing financial interest.

## Acknowledgements

This research is supported by National Natural Science Foundation of China (91226202), Ningbo Science and Technology Bureau (2014B82010) and Excellent Youth Foundation of Zhejiang Province of China (LR14B040001).

## Notes and references

- 1 X. Huang, Z. Y. Yin, S. X. Wu, X. Y. Qi, Q. Y. He, Q. C. Zhang, Q. Y. Yan, F. Boey and H. Zhang, *Small*, 2011, **7**, 1876–1902.
- 2 L. Lindsay and D. A. Broido, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **84**, 155421.
- 3 Y. Lin and J. W. Connell, *Nanoscale*, 2012, **4**, 6908–6939.
- 4 G. Eda, H. Yamaguchi, D. Voiry, T. Fujita, M. W. Chen and M. Chhowalla, *Nano Lett.*, 2011, **11**, 5111–5116.
- 5 S. Balendhran, J. K. Deng, J. Z. Ou, S. Walia, J. Scott, J. S. Tang, K. L. Wang, M. R. Field, S. Russo, S. Zhuiykov, M. S. Strano, N. Medhekar, S. Sriram, M. Bhaskaran and K. Kalantar-Zadeh, *Adv. Mater.*, 2013, **25**, 109–114.
- 6 M. Naguib, M. Kurtoglu, V. Presser, J. Lu, J. J. Niu, M. Heon, L. Hultman, Y. Gogotsi and M. W. Barsoum, *Adv. Mater.*, 2011, **23**, 4248–4253.
- 7 M. W. Barsoum, *Prog. Solid State Chem.*, 2000, **28**, 201–281.
- 8 Z. M. Sun, *Int. Mater. Rev.*, 2011, **56**, 143–166.
- 9 P. Eklund, M. Beckers, U. Jansson, H. Hogberg and L. Hultman, *Thin Solid Films*, 2010, **518**, 1851–1878.
- 10 J. Halim, M. R. Lukatskaya, K. M. Cook, J. Lu, C. R. Smith, L. A. Naslund, S. J. May, L. Hultman, Y. Gogotsi, P. Eklund and M. W. Barsoum, *Chem. Mater.*, 2014, **26**, 2374–2381.
- 11 M. Ghidui, M. R. Lukatskaya, M. Q. Zhao, Y. Gogotsi and M. W. Barsoum, *Nature*, 2014, **516**, 78–81.
- 12 M. Naguib, J. Come, B. Dyatkin, V. Presser, P. L. Taberna, P. Simon, M. W. Barsoum and Y. Gogotsi, *Electrochem. Commun.*, 2012, **16**, 61–64.
- 13 M. Naguib, O. Mashtalir, J. Carle, V. Presser, J. Lu, L. Hultman, Y. Gogotsi and M. W. Barsoum, *ACS Nano*, 2012, **6**, 1322–1331.
- 14 M. Naguib, J. Halim, J. Lu, K. M. Cook, L. Hultman, Y. Gogotsi and M. W. Barsoum, *J. Am. Chem. Soc.*, 2013, **135**, 15966–15969.
- 15 M. Ghidui, M. Naguib, C. Shi, O. Mashtalir, L. M. Pan, B. Zhang, J. Yang, Y. Gogotsi, S. J. L. Billinge and M. W. Barsoum, *Chem. Commun.*, 2014, **50**, 9517–9520.
- 16 O. Mashtalir, M. Naguib, V. N. Mochalin, Y. Dall'Agnese, M. Heon, M. W. Barsoum and Y. Gogotsi, *Nat. Commun.*, 2013, **4**, 1716.
- 17 M. R. Lukatskaya, O. Mashtalir, C. E. Ren, Y. Dall'Agnese, P. Rozier, P. L. Taberna, M. Naguib, P. Simon, M. W. Barsoum and Y. Gogotsi, *Science*, 2013, **341**, 1502–1505.
- 18 X. H. Xie, S. G. Chen, W. Ding, Y. Nie and Z. D. Wei, *Chem. Commun.*, 2013, **49**, 10112–10114.
- 19 X. J. Li, G. Y. Fan and C. M. Zeng, *Int. J. Hydrogen Energy*, 2014, **39**, 14927–14934.
- 20 Q. M. Peng, J. X. Guo, Q. R. Zhang, J. Y. Xiang, B. Z. Liu, A. G. Zhou, R. P. Liu and Y. J. Tian, *J. Am. Chem. Soc.*, 2014, **136**, 4113–4116.
- 21 X. D. Zhang, J. G. Xu, H. Wang, J. J. Zhang, H. B. Yan, B. C. Pan, J. F. Zhou and Y. Xie, *Angew. Chem., Int. Ed.*, 2013, **52**, 4361–4365.
- 22 X. H. Wang and Y. C. Zhou, *J. Mater. Sci. Technol.*, 2010, **26**, 385–416.
- 23 E. N. Hoffman, D. W. Vinson, R. L. Sindelar, D. J. Tallman, G. Kohse and M. W. Barsoum, *Nucl. Eng. Des.*, 2012, **244**, 17–24.
- 24 P. Xiao, J. C. Gu, J. Chen, D. Han, J. W. Zhang, H. T. Cao, R. B. Xing, Y. C. Han, W. Q. Wang and T. Chen, *Chem. Commun.*, 2013, **49**, 11167–11169.
- 25 J. Chen, K. Chen, D. Y. Tong, Y. J. Huang, J. W. Zhang, J. M. Xue, Q. Huang and T. Chen, *Chem. Commun.*, 2015, **51**, 314–317.
- 26 W. L. Liu, *et al.*, unpublished work.
- 27 M. Steenackers, A. Kuller, S. Stoycheva, M. Grunze and R. Jordan, *Langmuir*, 2009, **25**, 2225–2231.
- 28 J. E. Spanier, S. Gupta, M. Amer and M. W. Barsoum, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2005, **71**, 012103.
- 29 I. R. Shein and A. L. Ivanovskii, *Comput. Mater. Sci.*, 2012, **65**, 104–114.
- 30 I. Amin, M. Steenackers, N. Zhang, A. Beyer, X. H. Zhang, T. Pirzer, T. Hugel, R. Jordan and A. Golzhauser, *Small*, 2010, **6**, 1623–1630.
- 31 Z. Q. Niu, W. Y. Zhou, J. Chen, G. X. Feng, H. Li, W. J. Ma, J. Z. Li, H. B. Dong, Y. Ren, D. A. Zhao and S. S. Xie, *Energy Environ. Sci.*, 2011, **4**, 1440–1446.
- 32 L. B. Liu, H. J. Kim and M. Lee, *Soft Matter*, 2011, **7**, 91–95.

- 33 W. J. Ma, L. Song, R. Yang, T. H. Zhang, Y. C. Zhao, L. F. Sun, Y. Ren, D. F. Liu, L. F. Liu, J. Shen, Z. X. Zhang, Y. J. Xiang, W. Y. Zhou and S. S. Xie, *Nano Lett.*, 2007, 7, 2307–2311.
- 34 B. Zhao and W. J. Brittain, *Prog. Polym. Sci.*, 2000, 25, 677–710.
- 35 T. Chen, I. Amin and R. Jordan, *Chem. Soc. Rev.*, 2012, 41, 3280–3296.
- 36 J. Chen, P. Xiao, J. C. Gu, Y. J. Huang, J. W. Zhang, W. Q. Wang and T. Chen, *RSC Adv.*, 2014, 4, 44480–44485.

www.spm.com.cn